Synthesis of Spiro[indoline-3,2'-pyrrolidine] Derivatives from β -3-Indolyl Ketone Oximes

Kenichi Tanaka, Yutaka Mori, and Koichi Narasaka*

Department of Chemistry, Graduate School of Science, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-0033

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Spiro[indoline-3,2'-pyrrolidine] derivatives were prepared from β -3-indolyl ketone oximes by treatment with methanesulfonyl chloride and triethylamine via the bond formation between 3-position of the indole ring and oxime nitrogen. These diazaspirocycles were readily transformed to the corresponding spiro oxindole derivatives.

Spiro[indoline-3,2'-pyrrolidine] derivatives have received much attention in the development of pharmaceutically active molecules. These diazaspirocyles are mostly synthesized by the 1,3-dipoloar cycloaddition of azomethine ylides derived from isatins and secondary amines or α -amino acids.¹ Because the dipolarophiles available for the cycloaddition are limited to the activated alkenes, the alternative methods need to be developed for the preparation of a wide variety of the azaspiro indolines. Although the cyclization of 3-(3-aminoalkyl)indoles would be a candidate, there has been reported only one example of such a cyclization by the oxidation of 3-(3-ethylaminopropyl)indole, giving the low yield of the cyclized product.²

Recently, we have developed the intramolecular S_N2 -type reaction at the oxime sp² nitrogen to prepare azaheterocycles.³ For the latest instances, cyclic imines and quinolines are synthesized form γ , δ -unsaturated and β -aryl ketone oximes.^{3g,h} Since the reaction of indoles and electrophiles normally takes place at their 3-position,⁴ we expected that spiro[indoline-3,2'-pyrrolidine]s would be obtained from β -3-indolyl ketone oximes by a similar intramolecular S_N2 -type cyclization. In this paper is described this transformation reaction briefly.

 β -3-Indolyl ketone oximes were readily prepared by the In-Cl₃-catalyzed reaction of indole and enones,^{4c} followed by the oximation⁸ of the resulting ketones. The syn and anti isomers⁵ of the oximes are separated by column chromatography on silica gel. Thus obtained 4-(3-indolyl)butan-2-one anti-oxime 1a was treated with methanesulfonyl chloride (1.1 molar amounts) and triethylamine (2.2 molar amounts) at 0 °C (Scheme 1). A complex reaction mixture was obtained and the desired indolenine derivative 3a was not isolated, whereas the same reaction of the syn isomer of **1a** gave only the O-sulfonylated oxime in 95% yield. As the intramolecular cyclization product, spiro indolenine 3a, might have been readily hydrolyzed, the isolation of the cyclized product was attempted after the protection of an imino group. When pentafluorobenzoyl chloride⁶ was added to the reaction mixture, spirocyclic 2-chloroindoline $4a^7$ was obtained in 79% yield. In addition, quenching the acylation mixture with aqueous sodium bicarbonate in acetone for 20 h yielded the corresponding 2-hydroxy derivative 5a.

The formation of the diazaspiro compounds 4a and 5a indicates that the indole ring attacks the oxime nitrogen at the 3-position, affording unstable indolenine derivative 3a. 4a was obtained by the *N*-pentafluorobenzoylation of 3a at the less hindered aldimine nitrogen.



Scheme 1. Synthesis of diazaspirocycle 4a and 5a from *anti*oxime 1a.^a

Some examples for the transformation of β -3-indolyl ketone oxime to spiro[indoline-3,2'-pyrrolidine] derivatives are shown in Table 1. Oximes **1b** and **1c** having alkyl substituents at the α - or β -position gave the 2-hydroxy cyclized products⁷ **5b** and **5c** in moderate yields, respectively (Entries 2 and 3).

Table 1. Transformation of 4-(3-indolyl)butan-2-one oximes 1 to spiro-cyclic 2-hydroxyindolines 5^{a}

	$ \begin{array}{c} $		Cl, Et ₃ N C, 1 h 2Cl ₂		R ³ R ² R ¹ Me
1 1			5 ₅ COCl, r HCO ₃ (aq tone, rt, 2	t, 4 h .) 20 h	N ² OH OC ₆ F ₅ 5
Entry	Oxime	\mathbb{R}^1	\mathbb{R}^2	R ³	Product (Yield)
1 ^b	1a	Н	Η	Н	5a (63%)
2^{c}	1b	Me	Η	Η	5b (72%)
3 ^b	1c	Η	Me	Me	5c (74%)

a $1:MsCl:Et_3N:C_6F_5COCl = 1:1.1:2.2:1.05.$

b NaHCO₃ (aq.) was added 4 h after the addition of C₆F₅COCl.

c NaHCO₃ (aq.) was added after the extraction of the product obtained by the treatment of C₆F₅COCI.

When β -(2-methyl-3-indolyl) ketone oxime **1d** was treated with methanesulfonyl chloride and triethylamine, the cyclization also proceeded smoothly to yield azaspirocyclic indolenine **3d**, which was stable enough to be isolated in 95% yield by a column chromatography on Florisil (Table 2, Entry 1). Also in a similar cyclization of oxime **1e** having a *tert*-butoxycarbonyl group at the β -position of the oxime, 68% yield of the corresponding bisimine **3e**⁷ was obtained (Table 2, Entry 2).

 Table 2. Conversion of 4-(3-indolyl)butan-2-one oximes 1 to spiroindolenine 3



a $1:MsCl:Et_3N = 1:1.1:2.2$.

b Isolated by column chromatography on Florisil.

c NMR yield (anthracene as an internal standard)

Spiro[pyrrolidine-2,3'-oxindole] derivatives are reported to have a antimicrobial and antifungal activity, ^{lj} and were readily obtained from diazaspirocycles **5**. As shown in Scheme 2, treatment of **5a** with Dess–Martin periodinane and the successive removal of *N*-pentafluorobenzoyl group resulted in the formation of oxindole **6a**. In addition, it is noteworthy that **6a** was synthesized from oxime **1a** in one-pot procedure. After oxime **1a** was treated with methanesulfonyl chloride and triethylamine, excess manganese dioxide was added to the mixture, yielding **6a** in 53% yield.



 $\label{eq:Reagents: (i) Dess-Martin periodinane, CH_2Cl_2; (ii) NaOMe, MeOH, rt, 10 h; (iii) MsCl, 2Et_3N, 0 \ ^{\circ}C, 1 h, CH_2Cl_2 then MnO_2 (15 molar amounts), rt, overnight.$

Scheme 2. Synthesis of spiro[pyrrolidine-2,3'-oxindole] derivative 6a from 1a or 5a.

General experimental procedure is as follows (Scheme 1): To a dichloromethane (3 mL) solution of 4-(3-indolyl)butan-2one oxime (**1a**) (0.30 mmol) and triethylamine (0.65 mmol) was added methanesulfonyl chloride (0.33 mmol) at 0 °C. After 1 h at 0 °C, the reaction mixture was treated with a dichloromethane (2 mL) solution of pentafluorobenzoyl chloride (0.31 mmol) and was stirred at room temperature for 4 h, followed by the addition of aqueous sodium bicarbonate and acetone. After 20 h with stirring, the organic materials were extracted with dichloromethane and dried over sodium sulfate. The solvent was removed in vacuo, and the crude products were purified by thin-layer chromatography (silica gel deactivated with triethylamine, hexane:ethyl acetate = 1:2) to afford 3',4'-dihydro-2-hydroxy-5'-methyl-1-pentafluorobenzoylspiro[indoline-3,2'-[2H]pyrrole] (**5a**) (75 mg, 63%). This work was supported by the Grant-in-Aid for Scientific Research on Priority Areas (A) No. 412 "Exploitation of Multi-Element Cyclic Molecules" and the Grant-in-Aid for The 21st Century COE Program for Frontiers in Fundamental Chemistry from the Ministry of Education, Culture, Sports, Science and Technology, Japan.

References and Notes

- a) H. Aridill, R. Grigg, V. Sridharan, S. Surendrakumar, S. 1 Thianpatangul, and S. Kanajun, J. Chem. Soc., Chem. Commun., 1986, 602. b) H. Ardill, M. J. R. Dorrity, R. Grigg, M.-S. Leon-Ming, J. F. Malone, V. Sridharan, and S. Thianpatanagul, Tetrahedron, 46, 6433 (1990). c) T. Coulter, R. Grigg, J. F. Malone, and V. Sridharan, Tetrahedron Lett., 32, 5417 (1991). d) D. Fokas, W. J. Ryan, D. S. Casebier, and D. L. Coffen, Tetrahedron Lett., 39, 2235 (1998). e) V. Nair, K. C. Sheela, N. P. Path, and G. K. Eigendorf, Tetrahedron Lett., 41, 6217 (2000). f) V. Nair, K. C. Sheela, and N. P. Rath, Chem. Lett., 2000, 980. g) A. A. Raj and R. Raghunathan, Tetrahedron, 57, 10293 (2001). h) S. El-Ahl Abel-Aziz, Heteroat. Chem., 13, 324 (2002). i) A. K. Ganguly, N. Seah, V. Popov, C. H. Wang, R. Kuang, A. K. Saksena, B. N. Prammanik, T. M. Chan, and A. T. McPhail, Tetrahedron Lett., 43, 8981 (2002). j) A. A. Raj, R. Raghunathan, M. R. SrideviKumari, and N. Raman, Bioorg. Med. Chem., 11, 407 (2003).
- 2 M. J. Kornet and A. P. Thio, J. Med. Chem., 19, 892 (1976).
- 3 a) H. Kusama, Y. Yamashita, and K. Narasaka, *Chem. Lett.*, 1995, 4. b) H. Kusama, K. Uchiyama, Y. Yamashita, and K. Narasaka, *Chem. Lett.*, 1995, 715. c) H. Kusama, Y. Yamashita, K. Uchiyama, and K. Narasaka, *Bull. Chem. Soc. Jpn.*, 70, 965 (1997). d) S. Mori, K. Uchiyama, Y. Hayashi, K. Narasaka, and E. Nakamura, *Chem. Lett.*, 1998, 111. e) K. Uchiyama, M. Yoshida, Y. Hayashi, and K. Narasaka, *Chem. Lett.*, 1998, 607. f) M. Yoshida, K. Uchiyama, and K. Narasaka, *Heterocycles*, 52, 681 (2000). g) M. Yoshida, M. Kitamura, and K. Narasaka, *Chem. Lett.*, 2002, 144. h) M. Kitamura, T. Kikuchi, M. Yoshida, and K. Narasaka, *Synthesis*, 2003, in press.
- For some recent examples, see: a) T. Okauchi, M. Itonaga, T. Minami, T. Owa, K. Kitoh, and H. Yoshino, *Org. Lett.*, 2, 1485 (2000). b) K. B. Jensen, J. Thorhauge, R. G. Hazell, and K. A. Jørgensen, *Angew. Chem., Int. Ed.*, 40, 160 (2001). c) J. S. Yadav, S. Abraham, B. V. S. Reddy, and G. Sabitha, *Synthesis*, 2001, 2165. d) M. Bandini, P. G. Cozzi, M. Giacomini, P. Melchiorre, S. Selva, and A. Umani-Ronchi, *J. Org. Chem.*, 67, 3700 (2002). e) D. A. Evans, K. A. Scheidt, K. R. Fandrick, H. W. Lam, and J. Wu, *J. Am. Chem. Soc.*, 125, 10780 (2003).
- 5 In this manuscript, syn and anti isomer of oxime derivatives are defined when the hydroxy group on the oxime nitrogen is oriented syn and anti to the nucleophilic moiety, respectively.
- 6 Trapping of the 3a with other electrophiles such as acetyl chloride, trifluoroacetic anhydride, benzyloxycarbonyl chloride, and *p*-toluenesulfonyl chloride gave the complex mixture of products.
- 7 Compound 4a, 5b, and 3e were obtained as a mixture of two diastereomers, while compound 5a and 5c were obtained as a single diastereomer. The stereochemistry of 5a-c has not been determined.
- 8 Yields of the oximes 1a-e: 1a, 95% (*anti:syn* = 3:1). 1b, 92% (*anti:syn* = 3:1). 1c, 96% (*anti:syn* = 4:1). 1d, 95% (*anti:syn* = 3:1). 1e, 95% (*anti:syn* = 7:2).